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“Electrochemical Behavior of Pt(IV) on Mercury Electrode in the Presence of Dimethylglyoxime”

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Page	Line	Error	Correction
179	3	bis-dimethylglyoximate	bis-dimethylglyoximate
179	Keywords	dime-thylglyoxime	dimethylglyoxime
181	20	...negative values ( $E_p = -1.055$ V)...	...negative values ( $E_p = -1.055$ V)...
183	Figure 2	... peak current(A) and peak potential(B)...	...peak potential(A) and peak current(B)...
185	Figure 4	Effect of the DMG concentration on peak height in...	Peak current dependence on preconcentration time in 0.1 M acetate buffer of pH 4.6, $1 \times 10^{-4}$ M DMG ; $\blacklozenge$ - 10 $\mu\text{g Pt/L}$ , $\blacksquare$ - 15 $\mu\text{g Pt/L}$ , $\times$ - 25 $\mu\text{g Pt/L}$ ; at -0.4 V; electrode area: 0.028 $\text{cm}^2$ .
187	Figure 6	Dependence of the peak current (A), peak potential (B), and (C), on the scan rate in 0.1 M acetate buffer of pH 4.6, $1 \times 10^{-4}$ M DMG + 15 $\mu\text{g Pt/L}$ ; deposition time: 2 min at -0.4 V; electrode area: 0.028 $\text{cm}^2$ .	Dependence of the peak current (A), peak potential (B), and (C), on the scan rate and $\ln(\text{scan rate})$ in 0.1 M acetate buffer of pH 4.6, $1 \times 10^{-4}$ M DMG + 15 $\mu\text{g Pt/L}$ ; deposition time: 2 min at -0.4 V; electrode area: 0.028 $\text{cm}^2$ . The same figure (D) shows the dependence of the peak current and peak potential at different scan rate; the other conditions are the same.

Figure 2 B (page 183) should be read as

