

# **RSM-Based Predictive Modelling of Gravimetric Corrosion Characteristics of Welded and Tempered UNS G10400 Carbon Steel in a Seawater Environment**

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Received 19/10/2023; accepted 07/02/2024

<https://doi.org/10.4152/pea.2025430405>

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## **Abstract**

In this study, gravimetric analysis was used to determine corrosion properties, i.e., WL and CR of welded and tempered UNS G10400 CS exposed to Sw. In order to determine optimal WL and CR, RSM approach was utilized to develop a prediction model. 20 runs of the experiment were used for statistical ANOVA. As response variables, the experimental design matrix included WL and CR. To optimize the results of GC method and develop a predictive regression model, CCD was utilized. By ANOVA, the generated models for WL and CR indicated a significant p-value (which is probability, under the null hypothesis assumption, of obtaining a result equal to or higher than the one observed) of  $< 0.0001$ . F-values and  $R^2$  showed good statistical correlation between experimental and predicted values. Optimal conditions for WL and CR occurred at 25 °C(as-welded), with an ET of 17 days, and their optimum values were 0.744 g and 0.769 mm/year, respectively. Validation of optimized predicted results showed an UE of 1.6 and 2.9%, for WL and CR, respectively, which was below 5%. This revealed that the generated model adequately predicted CP of welded and tempered UNS G10400 CS in a Sw environment.

**Keywords:** as-welded; GT; optimal conditions; predictive modeling; RSM; UNSG10400 CS.

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## **Introduction\***

As a direct result of industrialization and technical advancement, more than 90% of current construction equipment is made of CS [1]. CS is used in chemical processing, oil drilling and refining, pipelines, mining, construction and metal-processing equipment, despite its low corrosion resistance [2-4]. It is widely used due to its high strength, durability and cost-effectiveness [5-8]. Welding is a common method of joining different sections of CS pipes, structures and equipment, in manufacturing, marine, oil and gas industries [9-10].

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\*The abbreviations list is on page 265.

However, CS welding in industrial application requires stringent procedures, to ensure safety and structural integrity [11]. When exposed to aggressive media commonly found in Sw media, CS is vulnerable to corrosion [12-14]. Corrosion of CS welds in Sw is a common problem, due to the aggressive nature of this environment. The presence of chloride ions, high salinity and dissolved oxygen accelerates the corrosion process, leading to a weakened weld that can fail prematurely [15, 16]. Therefore, proper surface preparation, PWHT, coating, inspection and routine maintenance are essential to prevent corrosion and ensure long-term structural integrity [17, 18].

RSM is a statistical and experimental design method that uses a selected dependent variable and deviations in one or more independent or factor variants [19-24]. This technique helps to predict corrosion properties, if it is effectively modelled and optimized, as reported in literature [26, 39, 40]. RSM is used to determine the optimal value for dependent variables, by adjusting design parameters for the independent ones [27]. Using a range of experimental design software applications and methodologies, corrosion prediction models have been developed [28-31]. GC analysis, a basic and cost-effective analytical method [32, 33], can be used to determine WL and CR of metals and alloys exposed to aggressive media. After establishing the mass loss over time, this evaluation method may be employed to predict CR of the material in actual service. This work strived for identifying optimal CP and models for an effective corrosion study of welded CS exposed to a Sw environment, using RSM approach.

## Experimental methods

### *Data collection*

For this design optimization and modelling investigation, experiment data obtained from a corrosion study using GT were employed. The effectiveness of post-weld tempering (at 25, 550, 650 and 700 °C) in protecting CS welds from corrosion in a Sw environment was evaluated. As-welded and PWT samples (550, 650 and 700 °C) were weighed and completely immersed by suspension in a Sw test media contained in plastic containers, for different ET of 17, 34, 51, 68 and 85 days. The samples were taken out after a specific ET was reached. Corrosion products were removed by washing the samples with distilled water, in acetone, and drying them in ambient air. The coupons were then reweighed, to determine their final weights. Experimental data were collected and analyzed using Eqs. 1 and 2.

$$\Delta W = w_i - w_a \quad (1)$$

$$CR = \frac{8.76 \times 10^4 \times \Delta w}{\rho A t} \quad (2)$$

where  $\Delta W$  is change in mass (g),  $w_i$  and  $w_a$  represent the coupon's initial and final weights, respectively,  $\rho$  is density (7.85 g/cm<sup>3</sup>),  $A$  is the exposed surface area in cm<sup>2</sup> and  $t$  is time in h.

### *CCD*

According to GT results, post-weld tempering and ET had a significant impact on WL and CR of CS. A two-factor, two-level CCD [34] was used to compare the effects of post-weld tempering and ET on WL and CR of the alloy. Table 1 shows

the levels of the independent variables used for constructing experiments to determine corrosion properties. By conducting the experiments at random, errors were minimized. 20 runs were generated by the design matrix. Data statistical analysis was performed using Design Expert 11. Research works reported by [35-37] describe the fundamental steps of the optimization procedure, which include: conducting the statistical design experiments; determining the coefficient in a mathematical model; predicting the response and confirming the model adequacy, as set up in the experiment; and validating the predicting model.

**Table 1:** Experimental variables and levels for the corrosion properties of welded and tempered UNS G10400 CS in a Sw environment.

Parameters	Unit	Symbol	Levels	
			-1	1
T of PWT	°C	A	25	700
ET	days	B	17	85

**Results and discussion**

**RSM results**

RSM results are presented in Table 2. The interactive effects of post-weld tempering T and ET on WL and CR are shown based on the considered factors. These results suggest that the relationship between WL and CR decreased with increased post-weld tempering T and ET.

**Table 2:** RSM variables predicted responses data.

SD	Run	Factor 1	Factor 2	Response 1	Response 2
		A: T of PWT (°C)	B: ET (days)	WL (g)	CR (mm/yr)
15	1	650	85	0.0282	0.0245
11	2	25	34	0.7208	0.5489
13	3	650	51	0.298	0.1029
12	4	650	34	0.2901	0.1708
6	5	550	17	0.3321	0.3739
3	6	25	51	0.6829	0.3483
19	7	700	68	0.1508	0.0231
4	8	25	68	0.5098	0.1669
2	9	550	68	0.3145	0.1064
14	10	550	85	0.1183	0.0556
10	11	650	17	0.1871	0.2579
9	12	700	17	0.1066	0.195
8	13	550	51	0.4156	0.1764
20	14	700	85	0.0248	0.004
5	15	650	68	0.2107	0.0541
7	16	700	34	0.2165	0.1185
16	17	650	85	0.0282	0.0245
18	18	25	85	0.2416	0.0046
17	19	550	34	0.4214	0.2656
1	20	25	17	0.7435	0.7686

**ANOVA for WL and CR**

Statistical ANOVA is presented in Tables 3 and 4, respectively, which show the model’s F-values of 18.34 and 77.38, for WL and CR of CS, respectively. Results also indicate that the probability of an elevated F-value due to the system

SNR is less than 0.01%. Microstructural modifications in CS samples caused by welding may account for variations [33]. The probabilities of failure (prob > F) with values lower than 0.050 for both variables are shown in Tables 3 and 4. This outcome implies that A, B, AB, A<sup>2</sup> and B<sup>2</sup> generated model terms are significant. Predicted R<sup>2</sup> values for WL (0.8676) and CR (0.9651) exhibit significant agreement with their corresponding adjusted R<sup>2</sup> (0.8203 and 0.9526, respectively). The variation below 0.2 of R<sup>2</sup> indicates a high level of statistical agreement between actual and predicted values. The adequacy of precision was determined by evaluating SNR, which is seen as entirely acceptable, if higher than 4. Since WL and CR obtained a SNR of 14.2119 and 32.117, this strongly indicates the model's good performance, and its ability to effectively navigate the design space. Hence, it can be concluded that the quadratic regression model was effective in predicting WL and CR of CS, both in its as-welded and PWT states, when exposed to a Sw environment [33]. A polynomial quadratic regression model was created to establish the relationship between WL and CR of CS, which are shown in Eqs. 3 and 4.

$$WL = 0.578518 - 0.225819A - 0.158338B + 0.0926183AB - 0.121458A^2 - 0.190357B^2 \quad (3)$$

$$CR = 0.279154 - 0.14355A - 0.238758B + 0.143278AB - 0.0743834A^2 + 0.0382786B^2 \quad (4)$$

**Table 3:** Quadratic ANOVA model for WL of UNS G10400 CS.

Source	Sum of squares	DF	Mean square	F-value	p-value	
Model	0.9342	5	0.1868	18.34	< 0.0001	significant
T of A-PWT	0.5858	1	0.5858	57.52	< 0.0001	
ET of B	0.2098	1	0.2098	20.6	0.0005	
AB	0.0545	1	0.0545	5.35	0.0365	
A <sup>2</sup>	0.0214	1	0.0214	2.1	0.1696	
B <sup>2</sup>	0.1268	1	0.1268	12.45	0.0033	
Residual	0.1426	14	0.0102			
Total corr.	1.08	19				
SD	0.1009		R <sup>2</sup>	0.8676		
Mean	0.3118		Adj. R <sup>2</sup>	0.8203		
CV%	32.37		Pred. R <sup>2</sup>	0.6888		
			Adeq. precis.	14.2119		

**Table 4:** Quadratic model ANOVA for CR.

Source	Sum of squares	DF	Mean square	F-value	p-	
Model	0.731	5	0.1462	77.38	<	significant
T of A-PWT	0.2367	1	0.2367	125.3	<	
ET of B	0.477	1	0.477	252.49	<	
AB	0.1303	1	0.1303	68.97	<	
A <sup>2</sup>	0.008	1	0.008	4.24	0.0586	
B <sup>2</sup>	0.0051	1	0.0051	2.71	0.1217	
Residual	0.0264	14	0.0019			
Total corr.	0.7574	19				
SD	0.0435		R <sup>2</sup>	0.9651		
Mean	0.1914		Adj. R <sup>2</sup>	0.9526		
CV%	22.71		Pred. R <sup>2</sup>	0.8825		
			Adeq. precis.	32.117		

**Interactive analysis of CS**

Figs. 1 to 4 illustrate graphical representations of WL and CR pertaining to the vulnerability of welded and tempered CS samples in a Sw medium. The plots depicted in Figs. 1 and 2 exhibit a linear relationship between predicted and actual WL and CR. The data points exhibited a tendency to cluster in close

proximity to the best fit line. The results indicate that the chosen quadratic model was suitable for predicting WL and CR of CS in a marine environment.

Design-Expert® Software

Weight loss

Color points by value of Weight loss:

0.05  0.87

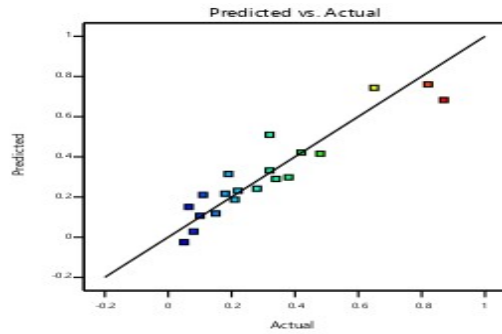


Figure 1: Predicted vs. actual WL.

Design-Expert® Software

Corrosion rate

Color points by value of Corrosion rate:

0.0124  0.804

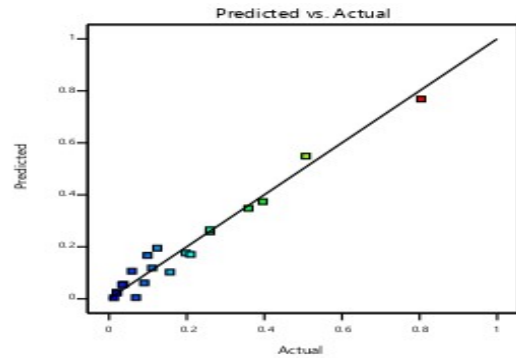


Figure 2: Predicted vs. actual CR.

Figs. 3 and 4 depict 3-D graphical representations of interactive effects resulting from the combination of post-weld tempering and ET.

Design-Expert® Software  
Factor Coding: Actual

Weight loss (g)

● Design points above predicted value

○ Design points below predicted value

0.05  0.87

X1 = A: PWT Temp  
X2 = B: Exposure Time

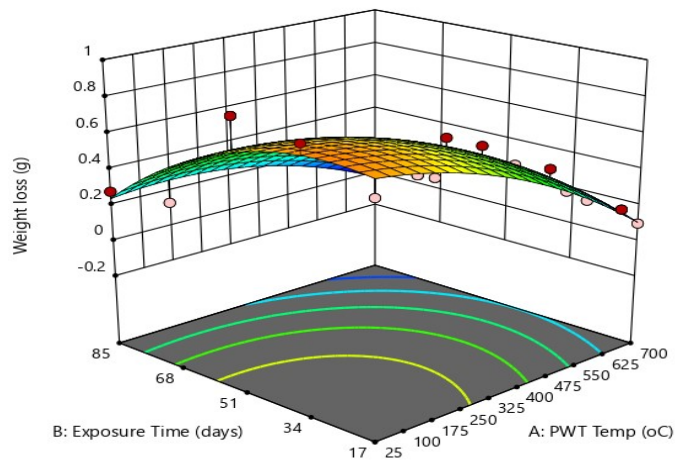
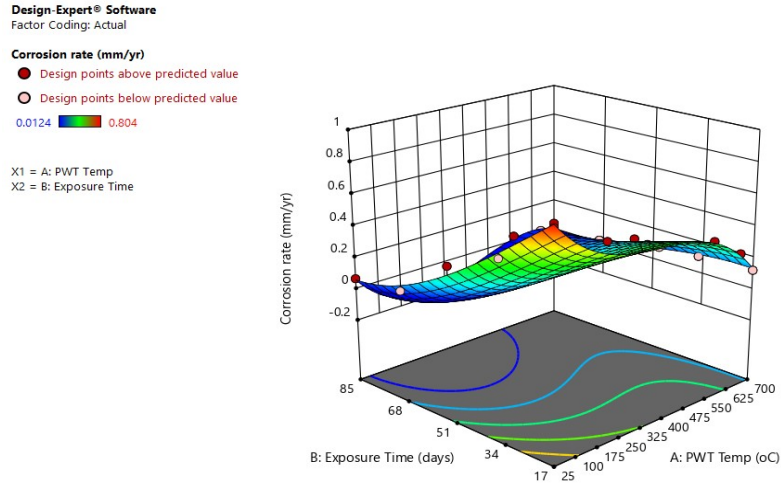


Figure 3: WL vs. ET and post-weld tempering T.

The study demonstrated a positive correlation between WL and CR, which has increased when the post-weld tempering T and ET decreased. CS metal dissolution retardation was due to the formation of a protective coating on its surface [38]. Optimal conditions predicted from RSM on the effect of post-weld tempering and ET of CS are 25 °C (as-welded) and 17 days of ET, and ideal WL and CR were predicted to be 0.744 g and 0.768 mm/yr, respectively. This shows that post-weld tempering is an effective method for improving corrosion properties of CS in a Sw environment.



**Figure 4:** CR vs. ET and post-weld tempering T.

### Validation of optimum parameters

To confirm the data validity, an additional experiment was conducted, as presented in Table 5. Experimental WL and CR values of 0.760 g and 0.798 mm/yr were close to the predicted values of 0.744 g and 0.769 mm/yr, respectively. UE for WL and CR were 1.6 and 2.9%. These values lower than 5% showed that the generated models adequately predicted WL and CR of welded and tempered CS in Sw.

**Table 5:** Validation of predicted data.

T of PWT (°C)	ET (days)	Experimental		Pred.		UE(%)	
		WL (g)	CR (mm/yr)	WL (g)	CR (mm/yr)	WL (g)	CR (mm/yr)
25	17	0.760	0.798	0.744	0.769	1.6	2.9

### Conclusion

The following conclusions can be deduced from this study. The optimal conditions for maximum WL and CR of UNS G10400 CS occurred at 25 °C (as-welded), with an ET of 17 days. Optimal values for WL and CR were 0.744 g and 0.769 mm/year, respectively. This shows that post-weld tempering was an effective method for improving corrosion properties of welded CS in a Sw environment.

ANOVA results of quadratic regression model used to fit the response revealed that it was significant ( $p < 0.0001$ ). F-values and  $R^2$  showed good statistical agreement between experimental and predicted values. This is an expression of the model degree of significance and fitness.

Validation of the optimized predicted results showed an UE of 1.6 and 2.9%, for WL and CR, respectively. This revealed that the generated model adequately predicted CP of welded and tempered CS in a Sw environment. The developed model can be effective in predicting appropriate output for welded and tempered UNS G10400 CS structures in a Sw environment.

### **Declaration of conflicting interests**

The authors declare that there is no potential conflict of interest with respect to the research, authorship, and/or publication of this article.

### **Authors' contributions**

**Benjamin U. Oreko, Silas O. Okuma:** conceptualized the research, drafted the manuscript, reviewed it and wrote its final version. **Silas O. Okuma:** carried out the experiments.

### **Abbreviations**

**ANOVA:** analysis of variance  
**CCD:** central composite design  
**CP:** corrosion parameters  
**CR:** corrosion rate  
**CS:** carbon steel  
**CV:** coefficient of variation  
**DF:** degree of freedom  
**ET:** exposure time  
**F:** failure  
**GC:** gravimetric corrosion  
**GT:** gravimetric techniques  
**PWHT:** post-weld heat treatment  
**PWT:** post-weld tempered  
 **$R^2$ :** coefficient of determination  
**RSM:** response surface methodology  
**SD:** standard deviation  
**SNR:** signal to noise ratio  
**Sw:** seawater  
**T:** temperature  
**UR:** uncertainty error  
**WL:** weight loss

### **References**

1. Valdez B, Schorr M, Cheng N et al. Technological applications of volatile corrosion inhibitors. *Corros Rev.* 2018;36(3):227. <http://dx.doi.org/10.1515/corrrev-2017-0102>

2. Bharatiya U, Gal P, Agrawal A et al. Effect of Corrosion on Crude Oil and Natural Gas Pipeline with Emphasis on Prevention by Ecofriendly Corrosion Inhibitors: A Comprehensive Review. *J Bio- and Tribo-Corros*. 2019;5(2):1-12. <http://dx.doi.org/10.1007/s40735-019-0225-9>
3. Valdez B, Ramirez J, Eliezer A et al. Corrosion assessment of infrastructure assets in coastal seas. *J Mater Eng Amp: Technol*. 2016;15(3):124-34. <http://dx.doi.org/10.1080/20464177.2016.1247635>
4. El-Meligi AA. Corrosion of Materials in Polluted Environment and Effect on World Economy. *Recent Path Corros Sci*. 2011;1(1):144-55. <http://dx.doi.org/10.2174/1877610811101020144>
5. Razvarz S, Jafari R, Gegov A et al. The Importance of Pipeline Transportation. *Stud Sys Deci and Cont*. 2020;2(1):1-24. <http://dx.doi.org/10.1007/978-3-030-59246-21>
6. Okuma SO, Orhorhoro EK, Tamuno RI et al. Corrosion evaluation on mild steel in different selected media. *Int J Eng Appl Sci Technol*. 2020;3(1):33-38. <http://doi.org/10.33564/IJEAST.2020.v05i03.006>
7. Okuma SO, Onyekwere KC. Inhibitive effect of *Irvingia gabonensis* leaf extract on the corrosion of mild steel in 1.0 M hydrochloric acid. *J Appl Sci Environ Manag*. 2022;26(1):25-29. <http://doi.org/10.4314/jasem.v26i1.4>
8. Okuma SO, Okerentie PO, Ihe GC et al. Investigating the effect of *Lasienthera africanum* extract as mild steel corrosion inhibitor in 0.5 M HCl solution. *Niger Res J Eng Environ Sci*. 2022;7(1):19-26.
9. Messler RW. *Principles of Welding*. 1999, Apr 23. <http://dx.doi.org/10.1002/9783527617487>
10. Easterling K. Fusion welding – process variables. *Introduction to the Physical Metallurgy of Welding*. 1992:1-5. <http://dx.doi.org/10.1016/b978-0-7506-0394-2.50006-x>
11. Phillips DH. *Welding engineering: an introduction*. John Wiley & Sons; 2023 Mar 21.
12. Ruiz-Cabañas FJ, Prieto C, Osuna R et al. Corrosion testing device for *in-situ* corrosion characterization in operational molten salts storage tanks: A516 Gr70 carbon steel performance under molten salts exposure. *Sol Ener Mater Sol Cells*. 2016;5(7):383-392. <http://dx.doi.org/10.1016/j.solmat.2016.06.005>
13. Xue W, Chen J, Xu F et al. Corrosion Development of Carbon Steel Grids and Shear Connectors in Cracked Composite Beams Exposed to Wet–Dry Cycles in Chloride Environment. *Mater*. 2018;11(4):479. <http://dx.doi.org/10.3390/ma11040479>
14. Verma J, Taiwade RV. Effect of welding processes and conditions on the microstructure, mechanical properties and corrosion resistance of duplex stainless steel weldments—A review. *J Manuf Pro*. 2017;2(5):134-52. <http://dx.doi.org/10.1016/j.jmapro.2016.11.003>
15. Stipaničev M, Turcu F, Esnault L et al. Corrosion behavior of carbon steel in presence of sulfate-reducing bacteria in seawater environment. *Electrochim Acta*. 2013;113:390-406. <http://dx.doi.org/10.1016/j.electacta.2013.09.059>
16. Pastorcic D, Vukelic G, Ivosevic S et al. Welded steel in marine environment – Experimental and numerical study of mechanical properties degradation. *Mater Today Comm*. 2023;34:105280. <http://dx.doi.org/10.1016/j.mtcomm.2022.105280>



17. Jiang J, Wang Y, Zhong Q et al. Preparation of Fe<sub>2</sub>B boride coating on low-carbon steel surfaces and its evaluation of hardness and corrosion resistance. *Surf Coat Technol.* 2011;(2)3:473-8. <http://dx.doi.org/10.1016/j.surfcoat.2011.07.053>
18. Gesnouin C, Hazarabedian A, Bruzzoni P et al. Effect of post-weld heat treatment on the microstructure and hydrogen permeation of 13CrNiMo steels. *Corros Sci.* 2004;46(7):1633-47. <http://dx.doi.org/10.1016/j.corsci.2003.10.006>
19. Goh KH, Lim TT, Chui PC et al. Evaluation of the effect of dosage, pH and contact time on high-dose phosphate inhibition for copper corrosion control using response surface methodology (RSM). *Corros Sci.* 2008;50(4):918-27. <http://dx.doi.org/10.1016/j.corsci.2007.12.008>
20. Chung NT, So YS, Kim WC et al. Evaluation of the Influence of the Combination of pH, Chloride, and Sulfate on the Corrosion Behavior of Pipeline Steel in Soil Using Response Surface Methodology. *Mater.* 2021;14(21):6596. <http://dx.doi.org/10.3390/ma14216596>
21. Sabour M, Dezvareh G, Bazzazzadeh R et al. Corrosion prediction using the weight loss model in the sewer pipes made from sulfur and cement concretes and Response Surface Methodology (RSM). *Constr Build Mater.* 2019;199:40-9. <http://dx.doi.org/10.1016/j.conbuildmat.2018.11.283>
22. Olawale O, Bello JO, Ogunsemi BT et al. Optimization of chicken nail extracts as corrosion inhibitor on mild steel in 2 M H<sub>2</sub>SO<sub>4</sub>. *Heliyon.* 2019;5(11):e02821. <http://dx.doi.org/10.1016/j.heliyon.2019.e02821>
23. Asmara YP, Athirah, Siregar JP et al. Application of response surface methodology method in designing corrosion inhibitor. *IOP Conf Ser: Mat Sci Eng.* 2017;257:012090. <http://dx.doi.org/10.1088/1757-899x/257/1/012090>
24. Chung NT, Choi SR, Kim JG et al. Comparison of Response Surface Methodologies and Artificial Neural Network Approaches to Predict the Corrosion Rate of Carbon Steel in Soil. *J Electrochem Soc.* 2022;169(5):051503. <http://dx.doi.org/10.1149/1945-7111/ac700d>
25. Salam K, Agarry S, Arinkoola A et al. Optimization of Operating Conditions Affecting Microbiologically Influenced Corrosion of Mild Steel Exposed to Crude Oil Environments Using Response Surface Methodology. *Brit Biotechnol J.* 2015;7(2):68-78. <http://dx.doi.org/10.9734/bbj/2015/16810>
26. Okewale AO, Omoruwuo F, Adesina OA et al. Comparative Studies of Response Surface Methodology (RSM) and Predictive Capacity of Artificial Neural Network (ANN) on Mild Steel Corrosion Inhibition using Water Hyacinth as an Inhibitor. *J Phy: Conf Ser.* 2019;1378(2):022002. <http://dx.doi.org/10.1088/1742-6596/1378/2/022002>
27. Gapsari F, Soenoko R, Suprpto A et al. Effect of Organics Corrosion Inhibitors on the Corrosion of 304SS in 3.5% NaCl. *Inter Rev Mech Eng (IREME).* 2016;(10)7:531. <http://dx.doi.org/10.15866/ireme.v10i7.9732>
28. Keshtegar B, Amine Ben Seghier M. Modified response surface method basis harmony search to predict the burst pressure of corroded pipelines. *Eng Fail Anal.* 2018;89(2):177-99. <http://dx.doi.org/10.1016/j.engfailanal.2018.02.016>

29. Asmara YP, Ismail MC. Efficient design of response surface experiment for corrosion prediction in CO<sub>2</sub> environments. *Corros Eng Sci Technol*. 2012;47(1):10-8. <http://dx.doi.org/10.1179/1743278211y.0000000013>
30. Liu J, Zhang T, Zhang W et al. Quantitative Modeling for Corrosion Behavior in Complex Coupled Environment by Response Surface Methodology. *Acta Metal Sin*. 2015;28(8):994-1001. <http://dx.doi.org/10.1007/s40195-015-0286-9>
31. Al-Sabur R. Tensile strength prediction of aluminium alloys welded by FSW using response surface methodology – Comparative review. *Mat Tod Proceed*. 2021;45:4504-10. <http://dx.doi.org/10.1016/j.matpr.2020.12.1001>
32. Edoziuno FO, Adediran AA, Odoni BU et al. Optimization and development of predictive models for the corrosion inhibition of mild steel in sulphuric acid by methyl-5-benzoyl-2-benzimidazole carbamate (mebendazole). *Cog Eng*. 2020;7(1):1714100. <http://dx.doi.org/10.1080/23311916.2020.1714100>
33. Edoziuno FO, Adediran AA, Adetunla A et al. RSM-based optimization and predictive modelling of the gravimetric corrosion behaviour of solution-treated copper-based shape memory alloy in HCl solution. *Inter J Interact Des Manuf. (IJIDeM)*. 2022;3(4):458-463. <http://dx.doi.org/10.1007/s12008-022-01163-x>
34. Ajeigbe SO, Aziz M, Basar N et al. Adsorption and Thermodynamic Characteristics of Phenylpropanoids of *Alpinia galanga* as Corrosion Inhibitors on Mild Steel. *Adv Sci Lett*. 2018;24(5):3561-3567. <http://dx.doi.org/10.1166/asl.2018.11436>
35. Saha P, Rao KVB. Biodegradation of commercial textile reactive dye mixtures by industrial effluent adapted bacterial consortium VITPBC6: a potential technique for treating textile effluents. *Biodegradation*. 2023;4(6):321-334. <http://dx.doi.org/10.1007/s10532-023-10047-0>
36. Nwaeju CC, Edoziuno FO, Adediran AA et al. Development of regression models to predict and optimize the composition and the mechanical properties of aluminium bronze alloy. *Adv Mat Proceed Technol*. 2021;8(3):1227-1244. <http://dx.doi.org/10.1080/2374068x.2021.1939556>
37. Nwaeju CC, Edoziuno FO, Nnuka EE et al. Predictive modeling and statistical analysis of mechanical properties OF heat treated Cu-10% Ni alloy using response surface methodology. *Mat Tod Proceed*. 2022;56(2):2371-2382. <http://dx.doi.org/10.1016/j.matpr.2021.12.167>
38. Olasehinde EF, Agbaffa BE, Adebayo MA et al. Corrosion protection of mild steel in acidic medium by titanium-based nanocomposite of *Chromolaena odorata* leaf extract. *Mater Chem Phys*. 2022;281:125856. <http://dx.doi.org/10.1016/j.matchemphys.2022.125856>
39. Umeozokwere AO, Mbabuike IU, Oreko BU et al. Corrosion Rates and its Impact on Mild Steel in Some Selected Environments. *J Sci Eng Res*. 2016;(3)1:34-43.
40. Oreko BU, Samuel B. Assessment of Inhibitive Drugs for Corrosion Inhibition Applications in Petrochemical Plants – A Review. *Saud J Eng Technol*. 2022;(7)5:201-210. <http://dx.doi.org/10.36348/sjet.2022.v07i05.001>